1	Calcium hydroxide solution can be added to magnesium chloride solution. The formula equation for the reaction is shown below:
	$MgCl_2(aq) + Ca(OH)_2(aq) \rightarrow Mg(OH)_2(s) + CaCl_2(aq)$
	This is an example of a precipitation reaction.
	What is the meaning of the term precipitation reaction?
	(2 marks)
1 (a) (i)	How could the magnesium hydroxide, $Mg(OH)_2$, be separated from the products of the reaction?
	(1 mark)
1 (b)	Barium chloride can react with sodium sulfate to produce barium sulfate and sodium chloride.
1 (b) (i)	Barium and chloride ions make barium chloride. In the space provided below, write the formula for barium chloride.
	Use the Chemistry data sheet to help you.
	+ Na ₂ SO4(aq) \rightarrow BaSO ₄ (s) + 2NaCl(aq)
	(1 mark)
1 (b) (i)	What does (aq) mean?
	(1 mark)
1 (c)	How would the barium sulfate be separated from the solution?
	(1 mark)
	(Total 6 marks)
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